DELHI PUBLIC SCHOOL, JAMMU

HALF YEARLY ASSIGNMENT(2019-20)

ASSIGNMENT : CHEMISTRY SUB: CHEMISTRY

CLASS:12TH

- Q.1. The rate constant for a first order reaction is $60s^{-1}$. How much time will it take to reduce the initial concentration of the reactant to its $1/16^{th}$ value?
- Q.2. Give four important difference between order of the reaction and molecularity of the reaction.
- Q.3. Define the following terms i) elementary reaction ii) half-life period of a reaction.
- Q.4 How is aspirin (Acetylsalicylic acid) prepared from salicylic acid?
- Q.5 Which out of propan-1-ol and propan-2-ol is stronger acid?
- Q.6 What is denaturation of an alcohol.
- Q.7 Give IUPAC name of CH₃OCH₂OCH₃
- Q.8 Diethyl ether does not react with sodium. Explain.
- Q.9 Under what condition do non –ideal solutions show negative deviations?
- Q.10 State Henry's law. What is the significance of K_H ?
- Q.11 Why is an increase in temperature observed on mixing chloroform and acetone?

 b) Why does sodium chloride solution freeze at a lower temperature than water?
- Q.12 Determine the amount of $CaCl_2$ (i= 2.47) dissolved in 2.5 litres of water such that its osmotic pressure is 0.75 atm at $27^{\circ}C$?
- Q.13 What do you mean by selectivity and activity of catalysts.
- Q.14 Adsorption is always exothermic. Why?
- Q.15 Chemisorption is highly specific. Illustrate with an example?
- Q.16 What is the difference between multimolecular and macromolecular colloids?
- Q.17 Differentiate between lyophilic and lyophobic colloids?

- Q.18 Give the working of fuel cell.
- Q.19 Give the mechanism for corrosion.
- Q.20 What is the role of silica in extraction of copper?
- Q.21 Give the working of zone of combustion in extraction of iron.
- Q.22 Alkyl halides are immiscible in water. Why?
- Q.23 How will you convert:
- (a) But-1-ene to But-2-ene
- (b) 1-Chlorobutane to 1-Flourobutane
- Q.24 Give the mechanism for the reaction of dimethyl ether with HI.
- Q.25 What is the role of depressants in froth floatation process?